

School of Arts, Science and Commerce

Department of Science

B.Sc Chemistry Semester II Major Assignment

Important Instructions to Student:

- 1. Last date for Assignment Submission **30-May-2020**
- 2. This assignment carries major **weightage of 50 Marks**. Kindly prepare it very carefully and in a very detailed manner. For any help in this regard, kindly contact your faculties.
- 3. Front Page of Assignment should clearly include details like:
 - a. Your Name
 - **b.** UID Number
 - c. Subject
 - **d.** Class
 - e. Semester

In the event of no such information, we may not be able to assign marks for your assignment, for which responsibility lies with students.

- 4. You can write and submit assignment through any of the following options:
 - a. Handwritten Assignment Prepare softcopy of your assignment through suitable apps and send the assignment as one PDF to your respective faculty as mentioned above.
 - b. Typed Assignment Prepare Assignment with following font setting and submit the assignment to your respective faculty as mentioned above.
 - i. Font Type Times New Roman or Arial
 - ii. Headings Font Size 14
 - iii. Text (Except Heading) 12
 - iv. Normal Margin and Line Spacing maximum 1.15
- 5. After this lockdown ends, you all have to submit the physical assignment copies to your respective Faculties. So, keep the assignment carefully for submission.
- 6. While submitting assignment through email, kindly use subject line as Name of the Programe_Name of Course/Branch_Semester_Name o the the Subject. For Example B.Tech._Mechanical_IV_Theory of Machines



| | eral chemistry . Nareshvari Chovatiya | Mode of Submission: email Email –nareshvari.chovatiya@raiuniversity.edu Subject Line: B.Sc sem-II ALL (General chemistry) | |
|---|---|---|--|
| 1. | Describe Heisenberg Uncertainty principle. | | |
| 2. | What is Hybridization? Explain in detail types of Hybridization. | | |
| 3 | Give the chemical and physical property of Noble gases. | | |
| 4 | Give the detail about VSEPR. | | |
| 5. | Define the term: electronegativity, ionization energy, Periodic trends. | | |
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| | eral Physics . Gayatri Sharma | Mode of Submission: Email Email—gayatri.sharma@raiuniversity.edu Subject Line: B.Sc. General Physics | |
| 1. | State and prove Gauss theorem in electrostatics. | | |
| 2. | Explain working principle of full wave rectifier with advantages and disadvantages. | | |
| 3 | State and explain joule Thomson effect with diagram. | | |
| 4 | Distinguish between an alternating current and direct current. | | |
| 5. | Derive the expression for interplanar spacing with diagram. | | |
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| English Communication & Samp; Life Skills Prof. Prof. Rakhi Pande | | Mode of Submission: email Email –rbs.raiuniversity@gmail.com Subject Line: B.Sc sem-II PCM | |
| 1. | Describe SQ3R study method in detail. | | |
| 2. | Mention all the helping verbs | | |
| 3 | Explain reading techniques in detail | | |
| 4 | Mention 10 habits that need to be developed | | |
| 5. | Explain the use of punctuation while using infinitive | | |
| Cub | icot. Numbou theory | Mode of Submission: Upload to link | |
| Subject: Number theory Faculty name: | | https://forms.gle/ic5rqR4HgSDj3UFV8 | |
| | dan Parmar | Subject Line: B.Sc sem-II PCM | |
| 1. | Prove that If $2^k - 1$ is prin | ne $(k > 1)$, then $n = 2^{k-1}(2^k - 1)$ is perfect number. | |
| 2. | Stat the Chinese Reminder Theorem and solve following system of linear congruence $x \equiv 2 \pmod{3}$ $x \equiv 3 \pmod{5}$ $x \equiv 2 \pmod{7}$ | | |
| 3 | Calculate 20! mod 23 using Wilson's Theorem. | | |
| 4 | Find the last two digits of 3 ²⁵⁶ mod100 | | |
| 5. | Explain with example hill cipher. | | |
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| EA Prof. | Dr. Sailesh Iyer | Mode of Submission: email Email — sailesh.iyer@raiuniversity.edu Subject Line: B.Sc sem-II PCM |
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| 1. | Explain types of e-Commerce. | |
| 2. | Explain MIS with appropriate examples. | |
| 3 | Explain Audio, Video and Animation in Multimedia with examples. | |
| 4 | Explain types of Operators in C. | |
| 5. | What is Traditional Business? Explain challenges in Traditional Business with examples. | |

NOTE: After completing your assignments, contact your respective faculty member and submit the assignment for assessment.